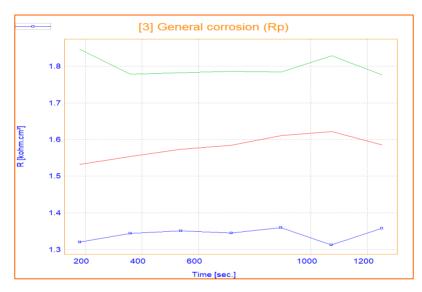
Application Note

# General Electrochemistry AP-C02

# General Corrosion (Rp)



This Application Note describes how the General Corrosion (Rp) method works by giving an example with Iron nail.





Application Note

## **Introduction**

The General corrosion (Rp) tests from OrigaMaster 5 are useful to investigate the efficiency of inhibitors used to prevent general (or uniform) corrosion. Automatic calculation of the resistance of polarization (called Rp) is determined from cyclic or linear voltammetries performed around the rest potential. The polarization resistance measures the instantaneous corrosion rate occurring at the electrochemical interface.

**NOTE:** The Rp can be used to evaluate the anti-corroding strength of an inhibitor.

# **Parameters**

Properties лх ۰ Display all 📃 Details ,14 Graph General corrosion (Rp) <-- Configure cell Start Scan rate (mV/sec.) before starting the experiment Overvoltage (mV) 40 OCP Duration (min.) 1 Open Circuit Determination (Rp) 7 Potential 🖉 Anodic Scan direction Smoothing 1 Maximum range Auto eral corrosion Minimum range Auto (Rp) Open circuit at end Yes 1 Save raw data Yes Calculation path All Equation GEC

The Parameter of the Rp test is shown in figure 1.

Figure 1: The parameters

With the above default settings, after 30 minutes of measuring OCP, individual voltammetries are recorded at 1mV/s.

The potential is scanned (in anodic direction, see page 2 to see all the available directions):

- from instantaneous <u>OCP</u> (Initial potential = free) to (OCP + 40 mV)
- and then back (in cathodic direction) to (OCP 40 mV).

By default, 7 Rp determinations are recorded. Between two successive measurements, the circuit is opened, and the cell is at rest (OCP) for 1 minute.

The circuit is opened at the end. The voltammetries are saved (if save raw data = Yes) as the rest potential versus time. Calculated Rp results (and corrosion rates) are saved versus time.



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# Parameters: Scan direction

Vg Imposed potential

This parameter sets the Potential scan. 4 scan directions are available.



Figure 2: Available scans

#### Anodic

The Potential is scanned to the Anodic part, then from there to the Cathodic part. So, by considering the parameters in the figure 1:

- First, from OCP
- to OCP + 40 mV
- Then, from OCP + 40 mV
- to OCP 40 mV
- Once again OCP, and as cvcles as Rp determinations.

#### Cathodic

The Potential is scanned to the Cathodic part, then from there to the Anodic part. So, by considering the parameters in the figure 1:

- First, from OCP
- to OCP 40 mV
- Then, from OCP 40 mV
- to OCP + 40 mV
- Once again OCP, and as cycles as Rp determinations.

#### None Anodic

The Potential is scanned from the Cathodic part to the Anodic part. So, by considering the parameters in the figure 1:

- Only, from OCP 40 mV
- to OCP + 40 mV
  - Once again OCP, and as cvcles as Rp determinations.



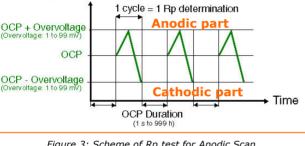


Figure 3: Scheme of Rp test for Anodic Scan

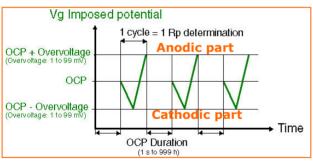


Figure 4: Scheme of Rp test for Cathodic Scan

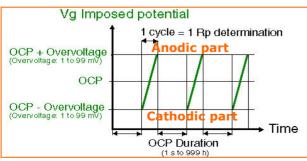


Figure 5: Scheme of Rp test for only Anodic direction

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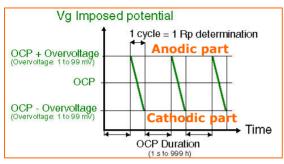


Figure 6: Scheme of Rp test for only Cathodic direction

#### **None Cathodic**

The Potential is scanned from the Anodic part to the Cathodic part. So, by considering the parameters in the figure 1:

- OCP + 40 mV to OCP 40 mV
- Once again OCP, and as cycles as Rp determinations.

### **Results**

Figure 7 shows the Rp values of 3 different tests, from different volume of Inhibitors.

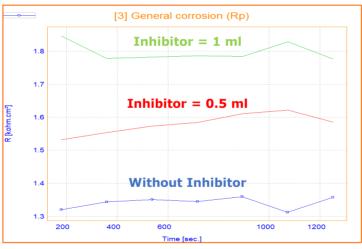


Figure 7: Overlay of Rp results from 3 different tests

The blue line = Rp measurements of Iron in NaCl 2 M - > without inhibitor.

The red line = Rp measurements of Iron in NaCl 2 M -> with 0.5 ml of inhibitor.

The green line = Rp measurements of Iron in NaCl 2 M -> with 1 ml of inhibitor

# **I**nterpretation



Figure 8: Overlay of 3 Rp results

# **Instrument and Electrodes**

By increasing the amount of inhibitors, we increase the Rp value too.

It means that the resistance of the metal increases. More the **Rp** is, less the corrosion is.

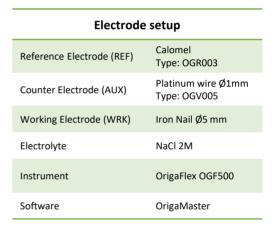
Using the **General Corrosion** method, we got the Rp value to study the quality of different kind of inhibitors.

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Figure 9: OrigaFlex OGF500



Figure 10: Electrochemical cell









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